PSEUDOGUAIANOLIDES AND CARABRONE DERIVATIVES FROM LOXOTHYSANUS SINUATUS

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Abstract—The aerial parts of Loxothysanus sinuatus afforded in addition to known compounds nine new pseudoguaianolides, three guaianolides and five carabrone derivatives. The structures were elucidated by spectroscopic methods. The chemotaxonomy is discussed briefly. The configurations of two guaianolides were corrected.

INTRODUCTION

The Mexican genus Loxothysanus was traditionally placed in the unnatural tribe Helenieae. The genera of this tribe have since been realigned into other tribes with which they might seem best related [1]. The position of Loxothysanus is unclear. We therefore have studied the constituents of L sinuatus (Less.) B L Robinson which may support a possible placement. The results will be discussed in this paper.

RESULTS AND DISCUSSION

The aerial parts of Loxothysanus sinuatus afforded germacrene D, caryophyllene, ent-kaurenic acid, its 15α -isobutyryloxy derivative, the pseudoguaianolides mexicanin [2], helenalin [3], helenalin tiglate [4], helenalin iso-valerate [4], mexicanin C [5], mexicanin C-tiglate [6], 6β -acetoxyamarilin [7], α -cyclocostunolide [8] and β -cyclocostunolide [9] as well as nine new pseudoguaianolides (1–9), the guaianolides 10–12 and the carabrone derivatives 13–17

The structure of 1 was deduced from the ¹H NMR spectrum (Table 1) and by NOE difference spectroscopy All signals could be assigned by spin decoupling and clear NOEs between H-15 and H-6, H-8 and H-10 as well as between H-14 and H-1, 2, 9α and 9β , between H-7 and H-9α and between 6-OH and H-13' established the stereochemistry at all chiral centres Accordingly, 1 is an 8epimer of helenalin (4) The ¹H NMR spectral data of 2 (Table 1) were close to those of 1 However, the presence of a tiglate residue caused the expected downfield shift of the H-6 signal and also a slight difference in the coupling $J_{6.7}$ indicating a small change in the conformation. The NOEs established that the configurations were still the same (H-15 with H-6, H-8 and H-10) As observed previously, the presence of 8a,12-olides also followed from the clear shift differences of H-8 in 1 and helenalin, its 1H NMR having been added for comparison in Table 1

The ¹H NMR spectra of 3–8 (Table 2) indicated that all compounds must be very similar, only those of 7 and 8 showed that an additional hydroxy group was present

Table 1 ¹H NMR spectral data of 1, 2 and helenalin (400 MHz, CDCl₃, TMS as internal standard)

-	1	2	Helenalın
H-1	2 98 ddd	3 08 ddd	3 05 ddd
H-2	7 76 dd	7 71 dd	7 68 dd
H-3	6 13 dd	6 08 dd	6 09 dd
H-6	4 14 dd	5 68 d	4 45 dd
H-7	2 91 dddd	3 12 dddd	3 55 br ddda
H-8	4 52 ddd	4 64 ddd	4 97 ddd
Η-9α	1 49 ddd	1 56 ddd	1 82 ddd
H-9ß	2 58 ddd	2 58 ddd	2 26 ddd
H-10	2 04 dddg	2 03 dddg	2 07 dddg
H-13	623 <i>d</i>	6 22 d	6 38 d
H-13'	6 01 d	5 97 d	5 78 d
H-14	1 24 d	1 28 d	1 25 d
H-15	1 15 s	1 20 s	0 90 s
OR		6 70 qq	
		1 73 br d	
OH	3 07 d	1 72 br s	2 32 d

J (Hz) 1, 2 = 2, 1, 3 = 3, 1, 10 = 11, 2, 3 = 6, 7, 8 = 10 5, 7, 13 = 7, 13' = 8, 9 β = 3, 8, 9 α = 11 5, 9 α , 9 β = 13, 9, 10 = 4 5, 9', 10 = 13, 10, 13 = 7, compound 1 6, 7 = 9, 6, OH = 2, compound 2 6, 7 = 7 5, helenalin 1, 2 = 2, 1, 3 = 3, 1, 10 = 12, 2, 3 = 6, 6, 7 = 1 5, 6, OH = 4, 7, 8 = 7 5, 7, 13 = 3, 7, 13' = 3, 8, 9 α = 2 5, 8, 9 β = 9, 9 α , 9 β = 15, 9 α , 10 = 6 5, 9 β , 10 = 4, 10, 14 = 7

Spin decoupling showed that this one had to be placed at C-6 Accordingly, a pair of double doublets in the spectra of 3-6, which could be assigned to H-6 by decoupling, were replaced by a new low field signal at $\delta 495$ In the spectrum of 7 this signal was also a double doublet, but in that of 8 a broadened doublet was observed In both cases sharp doublets were visible after deuterium exchange Additional oxygen functions were at C-2 and C-4, as followed from the result of spin decoupling Starting with the H-8 signal, which could be assigned by irradiation of H-7, the whole sequence could be established Although

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already the couplings observed indicated the stereochemistry at C-1, C-2, C-4, C-6 and C-8, final proof was necessary NOE difference spectroscopy clearly established the proposed configurations of all chiral centres Irradiation of H-15 gave clear NOEs of H-2, H-3 β , H-4, H-8 and H-10 Further NOEs were obtained between H-14, H-9 α and H-2, between H-7 and H-1 and between H-1 and H-7

In the ¹H NMR spectrum of 7 the typical signals of an isovalerate were visible, but in that of 8 these signals were replaced by an acetate singlet while all other signals were nearly identical

The spectra of 3 and 4 again differed only in the signals of the ester groups While 4 was a diacetate, 3 showed in addition to an acetate singlet the typical isovalerate signals All signals could be assigned by spin decoupling which clearly showed that the oxygen functions were again at C-2 and C-4. The couplings were identical with those of 7 and 8, thus indicating the same stereochemistry. The relative position of the ester groups in 3 could be established by acetylation of 5, where the position of the isovalerate residue could be deduced from the chemical shift of H-2

The ¹H NMR spectrum of 6 clearly showed that in this lactone the isovalerate residue was replaced by a tiglate as followed from the typical signals of this ester group

The ¹H NMR spectrum of **9** (Table 3) was close to that of 6β -acetoxyamarılın [7] However, the signal of H-6 was shifted up field We have tried to transform **9** to 6β -

acetoxyamarılın, but this was not successful In addition to the diacetate only the 4-O-acetate was obtained The position of the acetate groups easily could be deduced from the ¹H NMR spectra (Table 3)

The ¹H NMR spectrum of 10 (Table 4) was close to that of a guaianolide isolated from Helenium puberulum [10] The acetate residue, however, was replaced by an isovaleryloxy group Biogenetic considerations suggest that both compounds have the same stereochemistry at C-2 and C-4 and therefore we have again studied this point NOE difference spectroscopy showed that indeed both oxygen functions at C-2 and C-4 were α-orientated, as clear NOEs were observed between H-15 and H-2 and H-3β, between H-14 and H-2, between H-7 and H-1 as well as between H-10 and H-8 Therefore the configuration of puberolide [10] has to be corrected to 2α-acetoxy-4αhydroxy- 1α , 10β H-guaia-5, 11(13)-dien- 8α , 12-olide ¹H NMR spectrum of 11 (Table 4) indicated that an isomer of 10 was present. The low field H-6 signal was replaced by a pair of double doublets which collapsed to doublets on irradiation at 2 64 (H-7) All signals were close to those of a guaianolide from a Geigeria species [11] Again the acetate singlet was replaced by the signals of an isovalerate Careful inspection of the ¹H NMR spectral data showed that also here a 2 α -ester group at C-2 was more likely Thus the configuration of the lactone from Geigeria [11] should be changed to 2\alpha-acetoxy-4\alphahydroxy-10βH-guaia-1(5),11(13)-dien-8α,12-olide

The ¹H NMR spectrum of 12 (Table 4) was close to

	3	4	5	6	7	8
H-1	2 32 dd	2 32 dd	2 32 dd	2 39 dd	2 24 dd	2 24 dd
H-2	5 01 ddd	5 04 ddd	5 00 ddd	4 99 ddd	5 00 ddd	5 00 ddd
Η-3α	1 50 dd	1 52 dd	1 50 dd	1 56 dd	1 5 4 dd	1 56 <i>dd</i>
H-3β	2 69 ddd	2 70 ddd	2 64 ddd	2 66 ddd	2 63 ddd	2 62 ddd
H-4	478d	4 78 d	3 71 br dd	3 71 br dd	3 87 br dd	3 88 br d
Η-6α	2 39 dd	2 39 dd	285 m	2 88 m	4 95 br dd	495 br d
Н-6В	1 19 dd	1 19 dd	1 14 dd	1 16 dd	_	
H-7	2 89 m	289 m	287 m	290 m	3 01 dddd	3 00 dddd
H-8	4 12 ddd	4 17 ddd	4 18 dd	4 20 ddd	4 62 ddd	4 62 ddd
Η-9α	1 46 <i>ddd</i>	1 47 ddd	1 42 ddd	1 43 ddd	1 37 ddd	1 38 ddd
H-9 <i>β</i>	2 35 ddd	2 35 ddd	2 35 ddd	2 36 ddd	2 4 0 ddd	2 39 ddd
H-10	192 <i>m</i>	1 93 m	1 91 m	1 95 m	1 98 m	1 99 m
H-13	6 18 d	6 17 d	6 18 d	6 18 d	6 40 d	6 40 d
H-13'	5 40 d	5 40 d	5 48 d	5 48 d	5 61 d	5 60 d
H-14	1 03 d	1 03 d	1 01 d	1 01 d	1 02 d	1 01 d
H-15	097s	097s	0 88 s	0 9 0 s	087s	0 86 s
OR	2 16 d		2 18 d	6 85 qq	2 16 d	

211m

0.93d

Table 2 ¹H NMR spectral data of 3-8 (400 MHz, CDCl₃, TMS as internal standard)

J (Hz) 1,2 = 7,1,10 = 11,2,3 α = 2,2,3 β = 95,3 α ,3 β = 16,3 β ,4 = 45,7,8 = 9,7,13 = 7,13′ = 35,8,9 α = 12,8,9 β = 3,9 α ,9 β = 13,9 α ,10 = 125,9 β ,10 = 4,10,14 = 7,compounds 3–6 6 α ,6 β = 155,6 β ,7 α = 125,compound 6 4,OH = 5,compounds 7 and 8 6 α ,7 = 35,compound 7 4,OH = 55,6,OH = 3

179 br d

182 brs

195 br d

2 11 taa

1 98 br d 1 66 br d

094d

Table 3 ¹H NMR spectral data of 9 and the acetates (400 MHz, CDCl₃, TMS as internal standard)

2.10m

093d

208s

2 10 s 2 04 s

OAc

OH

	9	4-O-Acetate	Diacetate	
H-1 165 br d		1 82 d	1 80 d	
H-2	3 43 br d	} 3 47 brs	3 46 br d	
H-3	3 39 br d	34/078	3 34 d	
H-4	3 90 br s	4 86 s	4 80 br s	
H-6	4 30 br d	4 41 dd	5 53 d	
H-7	2 92 dddd	2 93 dddd	3 05 dddd	
H-8	4 63 ddd	4 58 ddd	4 57 ddd	
Η-9α	1 30 ddd	1 30 ddd	1 38 <i>ddd</i>	
Η-9β	2 49 ddd	2 49 ddd	2 51 ddd	
H-10	2 09 m	2 10 m	2 10 m	
H-13	6 38 d	6 31 d	6 22 d	
H-13'	5 55 d	5 53 d	5 41 d	
H-14	1 23 d	1 24 d	1 28 d	
H-15	1 15 s	1 07 s	1 15 s	
OAc	_	2 16 s	207 s	
OH	_	2 02 d		

J (Hz) 1, 10 = 11 5, 2, 3 = 3, 6, 7 = 4 5, 7, 8 = 9, 7, 13 = 7, 13' = 3 5, 8, 9α = 12 5, 8, 9β = 3, 9α , 9β = 13 5, 9α , 10 = 13, 9β , 10 = 5 5, 4-O-acetate 6, OH = 3 5

that of florilenalin [12] However, the acetate methyl singlet indicated the presence of florilenalin 2-O-acetate as the H-2 signal was significantly shifted to lower fields The couplings of H-1 and H-2 clearly showed that the

stereochemistry was that of florilenalin, where the absolute configuration was established by X-ray [12] Also the ¹³C NMR spectrum nicely agreed with the structure (see Experimental) The acetate 12 already was prepared from florilenalin, but no full ¹H NMR data were reported [12]

205s

The ¹H NMR spectral data of 17 (Table 5) clearly showed that a carabrone derivative was present as most signals were close to those of the cyclopropane derivative However, the methyl singlet of the carabrone spectrum was replaced by a pair of doublets at $\delta 4$ 10 and 3 89 An additional singlet at $\delta 2$ 14 (3H) showed that obviously an acetoxymethylene group was present All data therefore nicely agreed with the structure 17 which was also supported by the fragmentation in the mass spectrum After elimination of acetic acid a McLafferty reaction (loss of acetone) occurred. Furthermore a loss of methyl ethyl ketone can be observed These fragments are not visible in the MS of carabrone itself. The spectral data of the epimeric mixture of 15 and 16 (Table 5), which could not be separated by TLC or HPLC indicated that these lactones were corresponding 4-hydroxy derivatives of 17 Accordingly, the singlet of the methyl ketone was replaced by a doublet and a triplet quartet at 3 78 was due to the H-4 proton. While the chemical shifts of H-14 were nearly the same as that of 17, in the spectrum of 13 and 14 which also could not be separated, these doublets were shifted up field and the acetate signal was missing All data therefore showed that we were dealing with the epimeric diols 13 and 14

As the concentrations of the epimers in 13, 14, 15 and 16

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Table 4 ¹H NMR spectral data of 10-12 (400 MHz, CDCl₃, TMS as internal standard)

	10	11	12
H-1	2 58 br ddd		2 10 br d
I-2	5 05 ddd	5 57 br d	5 22 br da
Ι-3α	2 41 dd	2 37 dd	2 15 dd
i-3β	1 71 dd	1 88 <i>br d</i>	1 88 br d
I-5		_	2 15 br da
Ι-6α	} 5 99 dd	2 85 dd	2 07 br da
[-6β	3 99 aa	2 91 br dd	1 47 ddd
-7	3 38 br dddd	2 64 <i>ddddd</i>	3 20 m
-8	3 82 ddd	4 02 ddd	4 58 ddd
[-9α	1 69 ddd	1 65 ddd	2 67 br da
I-9 <i>β</i>	2 45 ddd	2 30 ddd	2 28 dd
-10	1 58 m	2 50 m	
I-13	6 26 d	6 23 d	6 25 d
-13'	5 68 d	5 57 d	5 64 d
[-14	} 1 05 d	} 1 20 d	5 01 br s
[-14'	\ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \	1 20 4	4 82 br s
-15	1 32 s	1 32 s	1 15 s
R	2 18 d	2 17 d	
	2 10 m	2 10 m	
	0 94 d	0 94 d	

J (Hz) Compound 10 1, 2 = 2, 3 α = 24, 1, 6 = 3, 1, 10 = 10; 2, 3 β = 9, 3 α , 3 β = 125, 6, 7 = 4, 7, 8 = 9, 7, 13 = 7, 13' = 35, 8, 9 α = 11, 8, 9 β = 9 β , 10 = 3, 9 α , 9 β = 135, 9 α , 10 = 135, 10, 14 = 7, compound 11 2, 3 α = 7, 3 α , 3 β = 15, 6 α , 6 β = 16, 6 α , 7 = 7, 13 = 7, 13' = 3, 6 β , 7 = 115, 7, 8 = 10, 8, 9 α = 12, 8, 9 β = 9 β , 10 = 3, 9 α , 9 β = 125, 9 α , 10 = 11, 10, 14 = 7, compound 12 1, 2 = 25, 1, 5 = 12, 2, 3 α = 55, 3 α , 3 β = 16, 5, 6 β = 12, 6 α , 6 β = 14, 6 α , 7 = 5, 6 β , 7 = 10; 7, 8 = 95, 7, 13 = 7, 13' = 25, 8, 9 α = 35, 8, 9 β = 115, 9 α , 9 β = 13

were different, the signals of the epimers could be assigned Inspection of models showed that the observed shift differences in the spectra of 13 and 14 may be explained if hydrogen bonds between the hydroxy groups would lead to fixed conformations. Most likely in both cases the methyl at C-4 was equatorial. In the main epimer a clear NOE between H-5 and H-14 and between H-14' and H-8 led to a conformation where a deshielding effect on H-9 α could be expected. This would indicate that the main compound was that with a 4-S-configuration

Loxothysanus is a genus of problematical relationships It has been related variously to such distantly related genera, among others, as Bahia [13], Perityle [14], Hymenothrix [15] and Hymenopappus [16] All of these have at one time or another been included in the classically constituted tribe Helenieae [1] In spite of recent subdivisions of the latter in which each of the above mentioned genera are placed in different subtribes [14], the chemistry of Loxothysanus does not support inclusion into any of the suggested groupings Rather the chemical data suggest a position in the subtribe Gaillardinae, which is a primitive element which links the mass of the classically circumscribed chaffless Helenieae to the tribe Heliantheae Faced with the chemical data presented here, one is inclined to agree with the comment of Turner [15], "Loxothysanus is an old relict group whose origin is not easily traced via extant relatives"

Table 5 ¹H NMR spectral data of 13-17 (400 MHz, CDCl₃, TMS as internal standard)

	13 and 14	15 and 16	17	
H-1 0 52 (0 62) m		0 69 m	0 67 ddd	
H-2))	1 66 ddt	
H-2'	165 145	1.60.140	1 57 ddt	
H-3	1 65-1 45 m	1 60–1 40 m	1	
1-3'	}	}	} 2 53 dd	
H-4	3 92 (3 81) tq	3 78 tq	, <u> </u>	
H-5	0.52(0.62)m	0 69 m	0 72 ddd	
I -6	2 34 <i>ddd</i>	2 34 ddd	2 32 ddd	
I-6′	0 96 ddd	0 97 ddd	0 93 ddd	
H-7	3 10 <i>dddd</i>	3 10 <i>ddddd</i>	3 07 dddda	
H-8	4 85 ddd	4 83 ddd	4 80 ddd	
H-9	285 (277) dd	2 62 dd	2 59 dd	
H-9'	0 90 (0 92) dd	0 99 dd	0 94 dd	
H-13	6 24 d	6 24 d	6 23 d	
H-13'	5 56 d	5 59 d	5 57 d	
H-14	3 62 (3 59) d	4 08 (4 04) d	4 10 d	
I-14'	3 38 (3 51) d	3 96 (3 94) d	3 89 d	
I -15	1 21 (1 19) d	1 15 <i>d</i>	207s	
R		2 08 s	2 14 s	

J (Hz) 5,6=6,7=7, 5,6'=9, 6,6'=145, 6',7=13, 7,8=85, 7,13=7,13'=25, 8,9=7, 8,9'=11, 9,9'=145, 14,14'=12, compounds 13-16, 3,4=4, 15=65, compound 17, 1,2=1,2'=2,3=2',3=7,1,5=5, 2,2'=145

EXPERIMENTAL

The air dried plant material (voucher Turner 15472, TEX, collected in Mexico) was worked-up in the usual fashion [17] The CC fractions of the extract of the aerial parts (400 g) were as follows 1 (petrol), 2 (Et₂O-petrol, 1 9), 3 (Et₂O-petrol, 1 4 and Et₂O-petrol, 1 1), 4 (Et₂O), 5 (Et₂O), 6 (Et₂O-MeOH, 9 1), 7 (Et₂O-MeOH, 4 1) and 8 (Et₂O-MeOH, 1 1) TLC (petrol) of fraction 1 gave 150 mg germacrene D and 5 mg caryophyllene and fraction 2 contained 850 mg ent-kaurenic acid Fraction 3 also contained acids and was therefore first treated with CH2N2 TLC (Et₂O-petrol, 3 7) gave 500 mg methyl-ent-kaurenoate and 500 mg of the 15α-isobutyrate TLC of fraction 4 (Et₂O-petrol, 7 3, three developments) gave 150 mg ent-kaurenic acid, 8 mg mexicanin C-tiglate, 10 mg 4 and a mixture which afforded by HPLC (MeOH-H₂O, 13 7) 15 mg 9 (R, 8 min) further a mixture which was separated again by HPLC (MeOH-H2O, 1 1) yielding 2 mg mexicanin C-tiglate (R_t 105 min), 5 mg helenalin (R_t 11 min) and 4 mg helenalin isovalerate (R, 13 min) TLC (Et₂O-petrol, 1 1) of fraction 5 gave four bands (5/1-5/4) 5/1 was pure mexicanin C-tiglate (2 mg) 5/2 was separated by HPLC (MeOH-H₂O, 1 1) affording 4 mg 4 (R, 10 5 min), 2 mg mexicanin C-tiglate (R, 11 min), 4 mg 2 (R, 12 5 min), a mixture which was purified by TLC (CH₂Cl₂-C₆H₆-Et₂O, 1 1 3) to give 2 mg 6 $(R_f 045)$ and 8 mg 5 $(R_t 23 \text{ min})$ 5/3 was separated by HPLC (MeOH-H₂O, 1 1) to give eight fractions (5/3/1-5/3/8) 5/3/1 contained 16 mg 1 (R_t 45 min), 5/3/2 contained 10 mg 6 β acetoxyamarılın (R_t 50 min), 5/3/4 13 mg α -cyclocostunolide $(R_t 88 \text{ min})$ 5/3/5 $(R_t 120 \text{ min})$ was separated by TLC $(CH_2Cl_2-C_6H_6-Et_2O, 1 \ 1 \ 3)$ to get 3 mg 2 $(R_f \ 0.58)$ and a band which by HPLC (MeOH- H_2O , 9 11) gave 4 mg β -cyclocostunolide (R_t 20 min) and 2 mg helenalin (R_t 23 min), 5/3/6 $(R_1 \ 13 \text{ min})$ gave by TLC $(CH_2Cl_2-C_6H_6-Et_2O, \ 1 \ 1 \ 3)$ 3 mg helenalin, 5/3/7 (R_t 13 5 min) gave by TLC $(CH_2Cl_2-C_6H_6-Et_2O, 1 1 3) 2 \text{ mg } 6 (R_f 0 5) \text{ and } 3 \text{ mg } 11$

 $(R_f \ 0.42)$ and 5/3/8 $(R_t \ 24.0 \ \text{min})$ gave 2.5 mg 10 5/4 contained 90 mg 17 $(R_f \ 0.15)$

Fraction 6 was separated by TLC (CH₂Cl₂-C₆H₆-Et₂O, 1 1 3) to give two bands The first one (R_f 0 55) contained 5 mg 17 and the second one (R_f 0 40) was separated by HPLC (MeOH-H₂O, 1 1) affording 10 mg 17 (R_t 8 2 min), 500 mg 15 and 16 (R_t 10 0 min) and a mixture (R_t 5 5 min) which was separated again by HPLC (MeOH-H₂O, 2 3) to give a fraction (R_t 10 0 min) which was treated by CH₂N₂ to remove impurities of acid material and then again separated by HPLC (MeOH-H₂O, 1 1) to give 2 mg mexicanin C (R_t 8 min)

HPLC of fraction 7 (MeOH-H₂O, 1 1) gave 12 mg mexicanin (R, 5.7 min), 60 mg 12 (R, 7.8 min) and 100 mg 15 and 16 (R, 10 min) TLC of fraction 8 (Et₂O-MeOH, 20 1) gave three bands (8/1-8/3) 8/1 $(R_f 0.75)$ was separated again by TLC (Et₂O-CHCl₃, 1 1, two developments) to give 100 mg 15 and 16 $(R_f 0.20)$ and 9 mg 7 $(R_f 0.08)$ 8/2 was separated by HPLC (MeOH-H₂O, 1 1) to give 13 mg mexicanin $(R_i.6 \text{ min})$, 12 mg 8 $(R_i.6.9 \text{ min})$, 40 mg 12 $(R_i.8 \text{ min})$ and 40 mg 15 and 16 $(R_i.5.0 \text{ min})$ and a fraction $(R_i.6.8 \text{ min})$ which by TLC (Et₂O-MeOH, 9 1, three developments) yielded 8 mg 8 $(R_f.0.80)$, 8 mg 13 and 14 $(R_f.0.73)$ and 40 mg 12 $(R_f.0.60)$

Known compounds were identified by comparison of the 400 MHz ¹H NMR spectra with those of authentic material, by co-TLC or by rigorous structure elucidation by ¹H NMR, MS and IR spectroscopy and by comparison with the data from literature All compounds were pure by TLC and HPLC as well as by ¹H NMR spectroscopy though some could not be induced to crystallize

8-Epihelenalin (1) Colourless crystals, mp 179°, IR $v_{\text{max}}^{\text{CCL}_4}$ cm⁻¹ 3600 (OH), 1780 (γ -lactone), 1705 (C=CC=O), MS m/z (rel int) 262 121 [M]⁺ (18) (calc for C₁₅H₁₈O₄ 262 121), 244 [M-H₂O]⁺ (6), 234 [M-CO]⁺ (5), 216 [244-CO]⁺ (4), 196 (12), 178 (22), 124 (100),

$$\left[\alpha\right]_{24^{\circ}}^{\lambda} = \frac{589 \quad 578 \quad 546 \quad 436 \,\mathrm{nm}}{-7 \quad -7 \quad -9 \quad -26} \ (c \ 0 \ 84, \ CHCl_3)$$

8-Epihelenalin tiglate (2) Colourless oil, IR $\nu_{\rm max}^{\rm CHCl_3}$ cm $^{-1}$ 1770 (y-lactone), 1710 (C=C-C=O, C=C-CO₂R), MS m/z (rel int) 344 162 [M] $^+$ (3) (calc for C₂₀H₂₄O₅ 344 162), 262 [M - O=C=C(Me)CH=CH₂] $^+$ (2), 244 [M - RCO₂H] $^+$ (17), 216 [244 - CO] $^+$ (7), 188 [216 - CO] $^+$ (10), 83 [C₄H₇CO] $^+$ (100), 55 [83 - CO] $^+$ (39)

4-O-Acetyl-pulchellin-2-O-isovalerate (3) Colourless oil, IR $v_{\text{CML}}^{\text{CCL}}$ cm $^{-1}$ 1780 (γ -lactone), 1750, 1250 (OAc), 1750 (CO₂R), MS m/z (rel int) 392 220 [M] $^+$ (1) (calc for C₂₂H₃₂O₆ 392 220), 290 [M - RCO₂H] $^+$ (5), 230 [290 - AcOH] $^+$ (49), 85 [C₄H₉CO] $^+$ (84), 57 [85 - CO] $^+$ (100),

$$\left[\alpha\right]_{24^{\circ}}^{\lambda} = \frac{589 \quad 578 \quad 546 \quad 436 \,\mathrm{nm}}{-31 \quad -31 \quad -36 \quad -64} \,(c \ 10, \, \mathrm{CHCl_3})$$

Pulchellin diacetate (4) Colourless oil, IR $v_{\text{max}}^{\text{CCl}_4}$ cm⁻¹ 1775 (y-lactone), 1740, 1240 (OAc), MS m/z (rel int) 350 173 [M]⁺ (1) (calc for $C_{19}H_{26}O_6$ 350 173), 290 [M – AcOH]⁺ (4), 248 [290 – ketene]⁺ (21), 230 [290 – AcOH]⁺ (100), 215 [230 – Me]⁺ (8), $[\alpha]_{D}^{\text{Cd}}$ - 12, CHCl₃, c 0 5

Pulchellin-2-O-isovalerate (5) Colourless oil, IR $v_{\text{max}}^{\text{CCl}_4}$ cm⁻¹ 3600 (OH), 1775 (γ-lactone), 1735 (CO₂R), MS m/z (rel int) 350 209 [M]⁺ (2 5) (calc for C₂₀H₃₀O₅ 350 209), 248 [M - RCO₂H]⁺ (32), 230 [248 - H₂O]⁺ (24), 85 [C₄H₉CO]⁺ (98), 57 [85 - CO]⁺ (100), [α] $_{D}^{24}$ - 13, CHCl₃, c 0 43 8 mg 5 in 0 1 ml Ac₂O were heated for 1 hr at 70° TLC (Et₂O-petrol, 2 1) gave 6 mg 3 (R_f 0 45), identical with the natural acetate (¹H NMR and co-TLC)

Pulchellin-2-O-tiglate (6) Colourless oil, IR v CCl₄ cm⁻¹ 3600

(OH), 1775 (y-lactone), 1710 (C=CCO₂R), MS m/z (rel int) 348 194 [M]⁺ (0.4) (calc for C₂₀H₂₈O₅ 348 194), 248 [M-RCO₂H]⁺ (12), 230 [248-H₂O]⁺ (8), 83 [C₄H₇CO]⁺ (100), 55 [83-CO]⁺ (44).

6β-Hydroxypulchellin-2-O-isovalerate (7) Colourless crystals, mp 127° $IR \nu_{max}^{CHCl_3} cm^{-1}$ 3620 (OH), 1770 (γ-lactone), 1720 (CO₂R), MS m/z (rel int) 366 204 [M]⁺ (2 5) (calc for C₂₀H₃₀O₆ 366 204), 348 [M-H₂O]⁺ (1 5), 264 [M-RCO₂H]⁺ (48), 246 [264-H₂O]⁺ (62), 85 [C₄H₉CO]⁺ (64), 57 [85-CO]⁺ (100), [α]_D²⁶ + 11, MeOH, c 0 47

 6β -Hydroxypulchellin-2-O-acetate (8) Colourless crystals, mp 87°, IR v_{max}^{KB} cm⁻¹ 3420 (OH), 1760 (γ-lactone), 1740 (OAc), MS m/z (rel int) 324 157 [M]⁺ (2) (calc for $C_{17}H_{24}O_6$ 324 157), 264 [M – AcOH]⁺ (20), 246 [264 – H₂O]⁺ (59), 228 [246 – H₂O]⁺ (10), 218 [246 – CO]⁺ (14), 203 [218 – Me]⁺ (12), 134 (56), 107 (52), 85 (100),

$$\left[\alpha\right]_{24^{\circ}}^{\lambda} = \frac{589 \quad 578 \quad 546 \quad 436 \,\mathrm{nm}}{+14 \ +16 \ +18 \ +26} \,(c \ 0.97, \,\mathrm{MeOH})$$

6β-Hydroxyamarılın (9) Colourless oıl, IR $\nu_{\rm max}^{\rm CHCl_3}$ cm $^{-1}$ 3600 (OH), 1770 (γ-lactone) 5 mg 9 was heated for 5 hr with 0 1 ml Ac₂O at 70° HPLC (MeOH-H₂O, 1 1) gave 3 mg 4-O-acetate (R, 7 min) and 2 mg diacetate (R, 11 5 min) 4-O-acetate colourless oil, IR $\nu_{\rm max}^{\rm CHCl_3}$ cm $^{-1}$ 3610 (OH), 1770 (γ-lactone), 1740 (OAc), MS (CI, isobutane) 323 [M + 1] $^+$ (100) (calc for C₁₇H₂₂O₆ + 1), 305 [323 - H₂O] $^+$ (12), diacetate colourless oil, IR $\nu_{\rm max}^{\rm CHCl_3}$ cm $^{-1}$ 1770 (γ-lactone), 1740, 1260 (OAc), MS m/z (rel int) 364 152 [M] $^+$ (9) (calc for C₁₉H₂₄O₇ 364 152), 30 [M - OAc] $^+$ (11), 304 [M - OHAc] $^+$ (6), 262 [304 - ketene] $^+$ (100), 244 [304 - HOAc] $^+$ (22),

$$[\alpha]_{24^{\circ}}^{\lambda} = \frac{589 \quad 578 \quad 546 \quad 436 \,\mathrm{nm}}{+11 \quad +14 \quad +15 \quad +25} (c \ 0 \ 11, \ MeOH)$$

 4α -Hydroxy- 2α -isovaleryloxy- 1α , 10β H-guaia-5,11(13)-dien- 8α ,12-olide (10) Colourless oil, IR $v_{\text{max}}^{\text{CCL}}$ cm⁻¹ 3600 (OH), 1775 (y-lactone), 1735 (CO₂R), MS m/z (rel int) 348 194 [M]⁺ (2) (calc for $C_{20}H_{28}O_5$ 348 194), 246 [M-RCO₂H]⁺ (38), 228 [246- H_2O]⁺ (24), 85 [C_4H_9CO]⁺ (69), 57 [85-CO] (100)

 $\begin{array}{l} 4\alpha\text{-}Hydroxy\text{-}2\alpha\text{-}isovaleryloxy\text{-}}10\beta\text{H-}guaia\text{-}}1(5),11(13)\text{-}dien-8\alpha,12\text{-}olide (11) Colourless oil, IR $\nu^{\text{CCl}}_{\text{max}}$ cm$^{-1}$ 3600 (OH), 1780 (y-lactone), 1740 (CO_2R), MS m/z (rel int) 348 194 [M]$^+ (25) (calc for C_{20}H$_{28}$O_5 348 194), 330 [M$-$H$_{2}$O]$^+ (1), 246 [M$-RCO_{2}$H]$^+ (100), 231 [246$-$Me]$^+ (12), 228 [246$-H_{2}$O]$^+ (8), 85 [C$_{4}$H$_{9}$CO]$^+ (34), 57 [85$-$CO]$^+ (45) \end{array}$

Florilenalin-2-O-acetate (12) Colourless crystals, mp 127°, IR $v_{\text{max}}^{\text{KBr}}$ cm⁻¹ 3420 (OH), 1770 (y-lactone), 1730 (OAc), MS m/z (rel int) 306 147 [M]⁺ (3 5) (calc for $C_{17}H_{22}O_5$ 306 147), 246 [M - AcOH]⁺ (42), 231 [246 - Me]⁺ (8), 228 [246 - H₂O]⁺ (24), 204 [246 - $C_{3}H_{6}$]⁺ (100), ¹³C NMR (CDCl₃, C-1-C-15) 51 5d, 80 4d, 49 0t, 78 3s, 53 1d, 29 7t, 42 1d, 73 9d, 39 5t, 140 0s, 139 2s, 169 8s, 122 6t, 115 5t, 25 2q (OAc 170 6s, 21 2q)

$$[\alpha]_{24^{\circ}}^{\lambda} = \frac{589 \quad 578 \quad 546 \quad 436 \text{ nm}}{+79 \quad +83 \quad +94 \quad +154} (c \quad 5 \quad 28, \text{ MeOH})$$

14-Hydroxy-4H- and 4-epi-H-carabrone (13 and 14) Colourless oil, IR $\nu_{\max}^{\rm CHCl}$, cm $^{-1}$ 3620 (OH), 1760 (γ -lactone), MS (CI, isobutane) m/z (rel int) 267 [M+1] $^+$ (5), 249 [267 - H₂O] $^+$ (100), 231 [249 - H₂O] $^+$ (25),

$$[\alpha]_{24^{\circ}}^{\lambda} = \frac{589 \quad 578 \quad 546 \quad 436 \text{ nm}}{+72 \quad +77 \quad +87 \quad +150} (c \ 0 \ 42, \text{ MeOH})$$

14-Acetoxy-4H- and 4-epi-H-carabrone (15 and 16) Colourless oil, IR $v_{\max}^{CCl_k}$ cm⁻¹ 3620 (OH), 1775 (y-lactone), 1750, 1245 (OAc), MS (CI, isobutane) m/z (rel int) 249 [M+1-AcOH]⁺

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(100), 231 $[249 - H_2O]^+$ (22),

$$\left[\alpha\right]_{24^{\circ}}^{\lambda} = \frac{589 \quad 578 \quad 546 \,\mathrm{nm}}{+89 \ +95 \ +108} (c \ 35 \ 1, \ \mathrm{MeOH})$$

14-Acetoxycarabrone (17) Colourless crystals, mp 83°, IR $v_{\rm col}^{\rm CCl_4}$ cm $^{-1}$ 1770 (γ -lactone), 1740, 1240 (OAc), 1720 (C=O), MS m/z (rel int) 306 147 [M] $^+$ (25) (calc for C₁₇H₂₂O₅ 306 147), 264 [M - ketene] $^+$ (12), 246 [M - AcOH] $^+$ (38), 188 [246 - Me₂CO, McLafferty] $^+$ (24), 164 [246 - MeCOEt] $^+$ (60), 85 (100)

$$[\alpha]_{24^{\circ}}^{\lambda} = \frac{589 \quad 578 \quad 546 \quad 436 \text{ nm}}{+79 \quad +83 \quad +93 \quad +153} (c \ 5 \ 5, \text{ CHCl}_3)$$

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REFERENCES

- 1 Turner, B L and Powell, A M (1977) The Biology and Chemistry of the Compositae (Heywood, V H, Harborne, J B and Turner, B L, eds), p 699 Academic Press, London
- 2 Dominguez, E and Romo, J (1963) Tetrahedron 19, 1415
- 3 Herz, W, Romo de Vivar, J, Romo, J and Viswanathan, N

- (1963) J Am Chem Soc 85, 19
- 4 Willuhn, G, Pottger, P M and Mathiesen, U (1983) Planta Med 49, 226
- 5 Herz, W, Romo de Vivar, J, Romo, J and Viswanathan, N (1963) Tetrahedron 19, 1359
- 6 Bohlmann, F, Zdero, C, King, R M and Robinson, H (1984) Justus Liebigs Ann Chem 503
- 7 Bohlmann, F, Misra, L N, Jakupovic, J, King, R M and Robinson, H (1985) Phytochemistry (in press)
- 8 Romo de Vivar, J and Jimenez, H (1965) Tetrahedron 21, 1741
- 9 Yoshioka, H., Renold, W., Fischer, N. H., Higo, A and Mabry, T. J. (1970) Phytochemistry 9, 823
- 10 Bohlmann, F and Jakupovic, J (1979) Phytochemistry 18,
- 11 Bohlmann, F, Zdero, C and Ahmed, M (1982) Phytochemistry 21, 1679
- 12 Lee, K. H., Ibuka, T., Kozuka, M., McPhail, A. T. and Onan, K. D. (1974) Tetrahedron Letters 2287
- 13 Rydberg, P A (1914) N Am Fl 34, 1
- 14 Turner, B L (1962) Brittoma 14, 101
- 15 Turner, B L (1974) Wrightia 5, 45
- 16 Robinson, H (1981) Smithsonian Contrib Botany 51, 78
- 17 Bohlmann, F, Zdero, C, King, R M and Robinson, H (1984) Phytochemistry 23, 1979